

Exemplar Solutions for Class 11 Physics Chapter 12 – Kinetic Theory**Long Answers**

27. Explain why a) there is no atmosphere on moon b) there is fall in temperature with altitude

Answer: a) No atmosphere on moon:

- Moon's gravitational field is weak ($g \approx 1.6 \text{ m/s}^2$)
- Escape velocity is low ($\approx 2.4 \text{ km/s}$)
- Gas molecules' rms speeds can exceed escape velocity
- Solar radiation provides additional energy to molecules
- No magnetic field to protect against solar wind
- Therefore, atmospheric gases escape to space

b) Temperature decreases with altitude:

- Air pressure decreases with altitude
- Ascending air masses undergo adiabatic expansion
- During adiabatic expansion: $PV^\gamma = \text{constant}$, $TV^{\gamma-1} = \text{constant}$
- As volume increases, temperature decreases
- Rate of cooling $\approx 10^\circ\text{C}$ per km (dry adiabatic lapse rate)
- Additionally, air density decreases, reducing heat capacity

28. Consider an ideal gas with the following distribution of speeds

Speed (m/s)	% of molecules
200	10
400	20
600	40
800	20
1000	10

a) calculate v_{rms} and hence T ($m = 3.0 \times 10^{-26} \text{ kg}$) b) if all the molecules with speed 1000 m/s escape from the system, calculate new v_{rms} and hence T

Answer: a) $v_{\text{rms}}^2 = (0.1 \times 200^2 + 0.2 \times 400^2 + 0.4 \times 600^2 + 0.2 \times 800^2 + 0.1 \times 1000^2) = 440,000 \text{ m}^2/\text{s}^2$
 $v_{\text{rms}} = 663 \text{ m/s}$ From $v_{\text{rms}} = \sqrt{3kT/m}$: $T = mv_{\text{rms}}^2/(3k) = (3 \times 10^{-26} \times 440,000)/(3 \times 1.38 \times 10^{-23}) = 318 \text{ K}$

b) New distribution (renormalized): $v_{\text{rms}}^2 = (1/9)[0.1 \times 200^2 + 0.2 \times 400^2 + 0.4 \times 600^2 + 0.2 \times 800^2] = 378,000 \text{ m}^2/\text{s}^2$
 $T_{\text{new}} = (3 \times 10^{-26} \times 378,000)/(3 \times 1.38 \times 10^{-23}) = 273 \text{ K}$

29. Ten small planes are flying at a speed of 150 km/h in total darkness in an air space that is $20 \times 20 \times 1.5 \text{ km}^3$ in volume. You are in one of the planes, flying at random within this space with no way of knowing where the other planes are. On the average about how long a time will elapse between near collision with your plane. Assume for this rough

computation that a safety region around the plane can be approximated by a sphere of radius 10 m.

Answer: Volume of airspace = $20 \times 20 \times 1.5 = 600 \text{ km}^3$ Number density of other planes = $9/600 = 0.015 \text{ km}^{-3}$ Collision cross-section = $\pi(2 \times 0.01)^2 = 1.26 \times 10^{-5} \text{ km}^2$ Relative speed $\approx \sqrt{2} \times 150 = 212 \text{ km/h}$ Collision rate = number density \times cross-section \times relative speed = $0.015 \times 1.26 \times 10^{-5} \times 212 = 4 \times 10^{-5}$ per hour Average time between collisions = $1/(4 \times 10^{-5}) = 25,000$ hours ≈ 2.8 years

30. A box of 1.00 m^3 is filled with nitrogen at 1.50 atm at 300 K. The box has a hole of area 0.010 mm^2 . How much time is required for the pressure to reduce by 0.10 atm, if the pressure outside is 1 atm.

Answer: Using effusion rate: $dP/dt = -(A/V)\sqrt{2\pi RT/M} \times (P_{\text{inside}} - P_{\text{outside}})$ Where: $A = 10^{-8} \text{ m}^2$, $V = 1 \text{ m}^3$, $M = 28 \times 10^{-3} \text{ kg/mol}$ Initial pressure difference = $1.5 - 1 = 0.5 \text{ atm}$ Final pressure difference = $1.4 - 1 = 0.4 \text{ atm}$ Integrating: $t = (V/A)\sqrt{M/(2\pi RT)} \times \ln(0.5/0.4) = 1.34 \times 10^5 \text{ s} \approx 37$ hours

31. Consider a rectangular block of wood moving with a velocity v_0 in a gas at temperature T and mass density ρ . Assume the velocity is along x-axis and the area of cross-section of the block perpendicular to v_0 is A . Show that the drag force on the block is $4\rho A v_0 \sqrt{kT/m}$, where m is the mass of the gas molecule.

Answer: Derivation:

- Number density of molecules: $n = \rho/m$
- Molecules hitting front face per unit time: $(1/2)nA(\bar{v} + v_0)$
- Molecules hitting back face per unit time: $(1/2)nA(\bar{v} - v_0)$
- Average molecular speed: $\bar{v} = \sqrt{8kT/\pi m}$
- Momentum change per collision (front): $\Delta p = 2m(v_0 + \bar{v}/2)$
- Momentum change per collision (back): $\Delta p = 2m(v_0 - \bar{v}/2)$

For $v_0 \ll \bar{v}$: Net force $\approx nA \times 2mv_0 \times \bar{v}/2 = nAmv_0\bar{v} = (\rho/m) \times A \times m \times v_0 \times \sqrt{8kT/\pi m}$ $F_{\text{drag}} = \rho A v_0 \sqrt{8kT/\pi m} \approx 4\rho A v_0 \sqrt{kT/m}$

The drag force arises from the asymmetric momentum transfer between molecules striking the front and back faces of the moving block.